**V-Semester** 

		Semiconductor Physics		
ourse Code		21BSP51	CIE Marks	50
Feaching Hours/Week (		2:2:0:1	SEE Marks	50
Total Hours of Pedagog	у	40	Total Marks	100
Credits		03	Exam Hours	03 Hours
Course Learning	•			
	ill develop a s	student to learn the fundation	mental physics of semic	conductor
material.	_			
• The course also	covers the wo	orking and applications fu	ndamentals semiconduc	ctor devices.
<b>Teaching-Learning P</b> These are sample Strate		al Instructions) eacher can use to accelera	te the attainment of the	various course ou
		ture methods various type		
		pted so that the delivered		
	ractical skills in			
		be arranged for students in		
e	0	group learning to improve	2	•
		very concept can be applied	ed to the real world. Thi	is helps the studen
understanding				
		nts for self-study.		1.1.1.1.1
		ing questions in the class		al thinking.
7. Inspire the st	idents towards	the studies by giving new	w ideas and examples.	
		Module-1		00 11
Band Theory of Solids:		aram Kronia pappy m	del Classification of	08 Hours.
		agram, Kronig penny mo gap, band structure of sen		
		ap, band structure of sen fajority and minority char		
		(derivation), law of mas		
nd Energy gap in an Int			s action, relation betw	
		oint presentation, Videos		
	-	<b>t:</b> band structure of semic	conductors	
- Sen Stu	-, component	Module-2		
				001
Electrical conductivity			a lavel and Eami lavel	08 hours
		semiconductor. Impuritie lerivation), Variation of		
		or, Minority carrier life		
•		nuity equations (derivation	•	
coefficient (derivation),			onj. man enteet, uetem	innation of fiall
	**			
0.01		r point presentation, Vide		
Self st	lay Compone	nt: Applications of Hall e	errect.	
		Module-3		
Fransistors:				08 Hours
		vorking, Transistor Curr		
		ase (CB), Common Co		
		aphical analysis, Current		
		nt, hybrid parameters Tr		
	nybrid equiva	alent of CE, Emitter follo	ower, impedance match	ing (qualitative)
Jumericals.		- *		
	-	oint presentation, Videos		
Self study	Component:	Junction Transistor.		
		Module-4		
Field Effect Transistor	(FET):			08 hours
	T, idea of c	channel formation, pinc		
		inda Daria constructio	on of MOSFET and	working, I-V
haracteristics. MOSFI	21. MOS D	Tode, Basic construction	in of moorbi and	
haracteristics. MOSFI haracteristics, enhancer		etion modes, Compliment		C.
haracteristics. MOSFI haracteristics, enhancer pecial Diodes	ment and deple	etion modes, Compliment	ary MOS (CMOS)	-
naracteristics. MOSFI naracteristics, enhancer <b>pecial Diodes</b> ener Diode - VI Cha	ment and deple		ary MOS (CMOS) ipper, Tunneling Effec	-

Self study Component: Zener Diode - VI Characteristics         Module-5         O8 hour         tor devices fabrication:       08 hour         tor device fabrication process: Oxidation, Diffusion, Ion implantation, Lithography, Thi         tor device fabrication process: Oxidation, Diffusion, Ion implantation, Lithography, Thi         tor device fabrication process: Oxidation, Diffusion, Ion implantation, Lithography, Thi         tor device fabrication, Videos         Practical Topics:         Self-study Component: Semiconductor device fabrication process.         come (Course Skill Set)
Ctor devices fabrication:       08 hour         tor device fabrication process: Oxidation, Diffusion, Ion implantation, Lithography, Thi         ion technique, Epitaxy, Examples: P-N junction device fabrication.         Chalk and talk, Power point presentation, Videos         Practical Topics:         Self-study Component: Semiconductor device fabrication process.         come (Course Skill Set)
tor device fabrication process: Oxidation, Diffusion, Ion implantation, Lithography, Thi ion technique, Epitaxy, Examples: P-N junction device fabrication. Chalk and talk, Power point presentation, Videos <b>Practical Topics:</b> Self-study Component: Semiconductor device fabrication process. come (Course Skill Set)
Practical Topics: Self-study Component: Semiconductor device fabrication process. come (Course Skill Set)
Self-study Component: Semiconductor device fabrication process. come (Course Skill Set)
Self-study Component: Semiconductor device fabrication process. come (Course Skill Set)
come (Course Skill Set)
of the course the student will be able to :
tinguish materials based on their band structure
derstand the electrical properties of semiconductors
ables to explain different types of transistors.
derstand the semiconducting devices and their applications.
• • • • • • • • • • • • • • • • • • • •
in knowledge of the various fabrication process of the semiconductor device
Details (both CIE and SEE)
inge of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is inimum passing mark for the CIE is 40% of the maximum marks (20 marks). A studer med to have satisfied the academic requirements and earned the credits allotted to each rse if the student secures not less than 35% (18 Marks out of 50) in the semester-en (SEE).
Internal Evaluation:
Fests each of <b>20 Marks (duration 01 hour</b> )
at the end of 5th week of the semester st at the end of the 10th week of the semester
at the end of the 15th week of the semester
at the end of the 15th week of the seniester
nents each of 10 Marks
nment at the end of 4th week of the semester
signment at the end of 9th week of the semester
assion/Seminar/quiz any one of three suitably planned to attain the COs and POs for <b>2</b> <b>ation 01 hours</b> ) I of the 13th week of the semester
three tests, two assignments, and quiz/seminar/group discussion will be out of 100 mark scaled down to 50 marks
s stressed CIE, the portion of the syllabus should not be common /repeated for any of th he CIE. Each method of CIE should have a different syllabus portion of the course). ds /question paper is designed to attain the different levels of Bloom's taxonomy as pe e defined for the course. nd Examination:
will be conducted by University as per the scheduled timetable, with common question be subject ( <b>duration 03 hours</b> ) ion paper will have ten questions. Each question is set for 20 marks.
Il be 2 questions from each module. Each of the two questions under a module (with f 3 sub-questions), <b>should have a mix of topics</b> under that module.

# Suggested Learning Resources:

- Text books:
  - 1. Modern Physics, R.Murugeshan & Er.K.Shivaprasath, S.Chand (2015).
  - 2. Semiconductor Device Fundamentals, <u>Robert F. Pierret</u>, Pearson education
  - 3. Solid state Physics, S. O. Pillai, New age International Publishers
  - 4. Hand Book of Electronics, Gupta and Kumar, Pragati Prakasan, Meerut

#### **Reference books:**

- 1. Physics for degree students, C L Arora and P. S. Hemme, S Chand Publications
- 2. Solid State Physics, N.W. Ashcroft and N.D. Mermin, 1976, Cengage Learning
- 3. Introduction to Solid State Physics, Charles Kittel, Wiley India Pvt. Ltd.

#### Web links and Video Lectures (e-Resources):

- 1. <u>https://yo</u>utu.be/k6ZxP9Yr02E
- 2. https://youtu.be/JA3sCmrv11M
- 3. https://youtu.be/mHAyQhz0ILE
- 4. https://youtu.be/N01BYteinzE
- 5. https://www.classcentral.com/course/swayam-fundamentals-of-electronic-device-fabrication-14080

#### Activity Based Learning (Suggested Activities in Class)/ Practical Based learning

https://youtu.be/c0fs-sNWmMM

https://nptel.ac.in

https://swayam.gov.in

https://vlab.amrita.edu

V-Semester

Course Code		Atomic and Molecular Phys		
		21BSP52	CIE Marks	50
	s/Week (L:T:P: S)	2:2:0:0	SEE Marks	50
Fotal Hours of	Pedagogy	40	Total Marks	100
Credits		03	Exam Hours	03 Hours
Course Obje				
		lucate students the fundament		
-		niques for the analysis of ator		ipies.
		ge to characterize atomic & n	-	
	onment, and infrastru	ing techniques those are us	serut in fields such a	as research, the
		n a physical, chemical & life	science laboratory	
- 030 di	nary treat teeninques I	n a physical, chemical & me	service raboratory	
Teaching-Le	arning Process (Ger	neral Instructions)		
		n teacher can use to accelerate	the attainment of the	various course ou
		lecture methods various types		
anim	nation films may be a	dopted so that the delivered le		
	ied and practical skill			
		y be arranged for students in		
		or group learning to improve t		
	rstanding level.	v every concept can be applied	a to the real world. The	is helps the studen
	port and guide the stu	dents for self-study		
		inking questions in the class,	which promotes critica	al thinking.
		rds the studies by giving new		U
		Module-1		
Pedagogy		-triplet separation for interacti	. 1	U U
		r point presentation, Videos <b>nt:</b> Spectra of alkali atoms		
		nt: Spectra of alkali atoms		
	Self study Compone	nt: Spectra of alkali atoms Module-2		00 L
Quantum stat	Self study Compone	nt: Spectra of alkali atoms Module-2 an atom:	ions for energy leve	08 hours
Quantum stat Electron spin,	Self study Compone es of an electron in a Stern-Gerlach exp	nt: Spectra of alkali atoms Module-2 an atom: periment, Relativistic correct		ls of hydrogen,
<b>Quantum stat</b> Electron spin, Hyperfine stru	Self study Compone es of an electron in a Stern-Gerlach exp acture and isotopic	nt: Spectra of alkali atoms Module-2 an atom:	Zeeman Effect, Expe	ls of hydrogen, rimental Study,
<b>Quantum stat</b> Electron spin, Hyperfine stru Quantum thery	Self study Compone es of an electron in a Stern-Gerlach exp acture and isotopic of Normal Zeeman	nt: Spectra of alkali atoms Module-2 an atom: periment, Relativistic correct shift, Lande Interval rule,	Zeeman Effect, Expe	ls of hydrogen, rimental Study,
Quantum stat Electron spin, Hyperfine stru Quantum thery width of spectr	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line	nt: Spectra of alkali atoms Module-2 an atom: periment, Relativistic correct shift, Lande Interval rule,	Zeeman Effect, Expe et Effect. Paschen Bac	ls of hydrogen, rimental Study,
Quantum stat Electron spin, Hyperfine stru Quantum thery vidth of spectr	Self study Compone es of an electron in a Stern-Gerlach exp acture and isotopic of Normal Zeeman al line Chalk and talk, Po	nt: Spectra of alkali atoms Module-2 an atom: periment, Relativistic correct shift, Lande Interval rule, 2 and anomalous Zeeman Effec	Zeeman Effect, Expe et Effect. Paschen Back	ls of hydrogen, primental Study, k & Stark effect,
Quantum stat Electron spin, Hyperfine stru Quantum thery width of spectr	Self study Compone es of an electron in a Stern-Gerlach exp acture and isotopic of Normal Zeeman al line Chalk and talk, Po	nt: Spectra of alkali atoms Module-2 an atom: beriment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Video onent: Quantum states of an e	Zeeman Effect, Expe et Effect. Paschen Back	ls of hydrogen, primental Study, k & Stark effect,
Quantum stat Electron spin, Hyperfine stru Quantum thery width of spectr Pedagogy	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compo	nt: Spectra of alkali atoms Module-2 an atom: periment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Video	Zeeman Effect, Expe et Effect. Paschen Back	ls of hydrogen, rrimental Study, k & Stark effect,
Quantum stat Electron spin, Hyperfine stru Quantum thery width of spectr Pedagogy Molecular Spe	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compo	nt: Spectra of alkali atoms Module-2 an atom: beriment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Video onent: Quantum states of an e	Zeeman Effect, Expe et Effect. Paschen Back os electron, Electron spin.	ls of hydrogen, rrimental Study, k & Stark effect, 08 hours
Quantum stat Electron spin, Hyperfine stru Quantum thery width of spectr Pedagogy Molecular Spe Types of mol	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compo ectroscopy: ecular spectroscopy,	nt: Spectra of alkali atoms Module-2 an atom: beriment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Video onent: Quantum states of an e Module-3	Zeeman Effect, Expert to Effect. Paschen Back os electron, Electron spin.	ls of hydrogen, rimental Study, k & Stark effect, 08 hours s of molecules,
Quantum stat Electron spin, Hyperfine stru Quantum thery vidth of spectr Pedagogy Molecular Spe Types of mol- quantisation of /ibrational an	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compo ectroscopy: ecular spectroscopy, Vibrational and rota d electronic spectra	nt: Spectra of alkali atoms Module-2 an atom: beriment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Video onent: Quantum states of an e Module-3 Applications, Discrete set tional energies, determination a of diatomic and polyato	Zeeman Effect, Expert et Effect. Paschen Back os lectron, Electron spin. of electronic energie of internuclear distan- omic molecules, Bor	Is of hydrogen, primental Study, k & Stark effect, 08 hours s of molecules, nees, Rotational,
Quantum stat Electron spin, Hyperfine stru Quantum thery width of spectr Pedagogy Molecular Spe Types of mol- quantisation of vibrational an	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compo ectroscopy: ecular spectroscopy, Vibrational and rota d electronic spectra	nt: Spectra of alkali atoms Module-2 an atom: beriment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Video onent: Quantum states of an e Module-3 Applications, Discrete set tional energies, determination	Zeeman Effect, Expert et Effect. Paschen Back os lectron, Electron spin. of electronic energie of internuclear distan- omic molecules, Bor	Is of hydrogen, primental Study, k & Stark effect, 08 hours s of molecules, nees, Rotational,
Quantum stat Electron spin, Hyperfine stru Quantum thery width of spectr Pedagogy Molecular Spe Types of mol quantisation of vibrational an approximation,	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compo ectroscopy: ecular spectroscopy, vibrational and rota d electronic spectra , Frank – Condon pri	nt: Spectra of alkali atoms Module-2 an atom: beriment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Videc onent: Quantum states of an e Module-3 applications, Discrete set tional energies, determination a of diatomic and polyato inciple and selection rules. Me	Zeeman Effect, Expert et Effect. Paschen Back os lectron, Electron spin. of electronic energie of internuclear distan- omic molecules, Bor	Is of hydrogen, primental Study, k & Stark effect, 08 hours s of molecules, nees, Rotational,
Quantum stat Electron spin, Hyperfine stru Quantum thery vidth of spectr Pedagogy Molecular Spe Types of mol- quantisation of ribrational an pproximation, Pedagogy	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compo ectroscopy: ecular spectroscopy, vibrational and rota d electronic spectra , Frank – Condon pri Chalk and talk, Powe	nt: Spectra of alkali atoms Module-2 an atom: beriment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Video ment: Quantum states of an e Module-3 applications, Discrete set tional energies, determination a of diatomic and polyate inciple and selection rules. Me r point presentation, Videos	Zeeman Effect, Expert et Effect. Paschen Back os electron, Electron spin. of electronic energie of of internuclear distan- omic molecules, Bor olecular hydrogen.	Is of hydrogen, primental Study, k & Stark effect, 08 hours s of molecules, nees, Rotational,
Quantum stat Electron spin, Hyperfine stru Quantum thery width of spectr Pedagogy Molecular Spe Types of mol quantisation of vibrational an approximation, Pedagogy	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compo ectroscopy: ecular spectroscopy, vibrational and rota d electronic spectra , Frank – Condon pri Chalk and talk, Powe	nt: Spectra of alkali atoms Module-2 an atom: periment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Video onent: Quantum states of an e Module-3 Applications, Discrete set tional energies, determination a of diatomic and polyato inciple and selection rules. More r point presentation, Videos nt: Types of molecular Spect	Zeeman Effect, Expert et Effect. Paschen Back os electron, Electron spin. of electronic energie of of internuclear distan- omic molecules, Bor olecular hydrogen.	Is of hydrogen, primental Study, k & Stark effect, 08 hours s of molecules, nees, Rotational,
Quantum stat         Electron spin,         Hyperfine stru         Quantum thery         width of spectr         Pedagogy         Molecular Spectr         Types of molequantisation of         vibrational an         approximation,         Pedagogy         Quantisation of         Yetagogy	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compone ectroscopy: ecular spectroscopy, vibrational and rota d electronic spectra , Frank – Condon pri Chalk and talk, Powe Self study Compone	nt: Spectra of alkali atoms Module-2 an atom: beriment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Video ment: Quantum states of an e Module-3 applications, Discrete set tional energies, determination a of diatomic and polyate inciple and selection rules. Me r point presentation, Videos	Zeeman Effect, Expert et Effect. Paschen Back os electron, Electron spin. of electronic energie of of internuclear distan- omic molecules, Bor olecular hydrogen.	ls of hydrogen, rrimental Study, k & Stark effect, 08 hours s of molecules, nces, Rotational, n Oppenheimer
Quantum stat         Electron spin,         Hyperfine stru         Quantum thery         width of spectr         Pedagogy         Molecular Spectr         Types of mol-         quantisation of         vibrational an         approximation,         Pedagogy         Raman Spectr	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compone ectroscopy: ecular spectroscopy, vibrational and rota d electronic spectra , Frank – Condon pri Chalk and talk, Powe Self study Compone	Int: Spectra of alkali atoms         Module-2         an atom:         periment, Relativistic correct         shift, Lande Interval rule, 1         and anomalous Zeeman Effect         wer point presentation, Video         oment: Quantum states of an e         Module-3         applications, Discrete set         tional energies, determination         a of diatomic and polyator         inciple and selection rules. Market         r point presentation, Videos         nt: Types of molecular Spect         Module-4	Zeeman Effect, Expert et Effect. Paschen Back os electron, Electron spin. of electronic energie n of internuclear distant omic molecules, Bor olecular hydrogen.	ls of hydrogen, rrimental Study, k & Stark effect, 08 hours s of molecules, nces, Rotational, n Oppenheimer 08 hours
Quantum stat         Electron spin,         Hyperfine stru         Quantum thery         width of spectr         Pedagogy         Molecular Spe         Types of mol-         quantisation of         vibrational an         approximation,         Pedagogy         Raman Spectr         Raman effect,	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compone ectroscopy: ecular spectroscopy, vibrational and rota d electronic spectra , Frank – Condon pri Chalk and talk, Powe Self study Compone roscopy: Rotational Raman sp	nt: Spectra of alkali atoms Module-2 an atom: periment, Relativistic correct shift, Lande Interval rule, 1 and anomalous Zeeman Effect wer point presentation, Video onent: Quantum states of an e Module-3 Applications, Discrete set tional energies, determination a of diatomic and polyato inciple and selection rules. More r point presentation, Videos nt: Types of molecular Spect	Zeeman Effect, Expert et Effect. Paschen Back os electron, Electron spin. of electronic energie n of internuclear distan- omic molecules, Bor olecular hydrogen. roscopy ectra. Stokes and anti-	Is of hydrogen, primental Study, k & Stark effect, 08 hours s of molecules, nces, Rotational, n Oppenheimer 08 hours Stokes lines and
Quantum state         Electron spin,         Hyperfine stru         Quantum thery         Vidth of spectre         Pedagogy         Molecular Spectre         Types of molecular in the second structure         Quantisation of vibrational an approximation.         Pedagogy         Raman Spectre         Raman effect,         heir Intensity	Self study Compone es of an electron in a Stern-Gerlach exp icture and isotopic of Normal Zeeman al line Chalk and talk, Po Self study Compo ectroscopy: ecular spectroscopy, vibrational and rota d electronic spectr; Frank – Condon pri Chalk and talk, Powe Self study Compone roscopy: Rotational Raman sp difference, Instrum	<b>Module-2 Module-2 an atom:</b> periment, Relativistic correct shift, Lande Interval rule, 1         and anomalous Zeeman Effect         wer point presentation, Video <b>Module-3</b> Applications, Discrete set         tional energies, determination         a of diatomic and polyato         inciple and selection rules. Market         r point presentation, Videos <b>nt:</b> Types of molecular Spect <b>Module-4</b> weetra. Vibrational Raman spectral	Zeeman Effect, Expert et Effect. Paschen Back os electron, Electron spin. of electronic energie n of internuclear distan- omic molecules, Bor olecular hydrogen. roscopy ectra. Stokes and anti- Fluorescence and P	Is of hydrogen, primental Study, k & Stark effect, 08 hours s of molecules, nces, Rotational, n Oppenheimer 08 hours Stokes lines and hosphorescence,

Pedagogy	Chalk and talk, Power point presentation, Videos
	Self study Component: Raman effect
	Module-5
Characterizatelements of Spectroscopy	<b>netic Spectrum:</b> 08 hours ation of Electromagnetic Radiation. regions of spectrums, spectra representation, basis practical spectroscopy, resolving power, width and intensity of spectral transition. NMI y: Nuclear spin, nuclear resonance, saturation, spin-spin and spin-lattice relaxation off, deshielding, coupling constant, instrumentation and applications.
Pedagogy	Chalk and talk, Power point presentation, Videos
	Self study Component: NMR applications
Course outc	come (Course Skill Set)
	of the course the student will be able to
• Stud	dy the particulars of spectra of Hydrogen-like atoms
• Ana	alyse the Zeeman effect
• Cor	mprehend the specifics of rotational, vibrational and Raman spectra of molecules.
• Gai	in knowledge of NMR spectra and their applications
• Rec	cognize the various regions of spectra and the conforming instrumentations
50%. The m shall be deer subject/ coun examination( <b>Continuous</b> Three Unit T 1. First test a	age of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is aninimum passing mark for the CIE is 40% of the maximum marks (20 marks). A studer emed to have satisfied the academic requirements and earned the credits allotted to eac rse if the student secures not less than 35% (18 Marks out of 50) in the semester-en (SEE). Internal Evaluation: Tests each of 20 Marks (duration 01 hour) at the end of 5th week of the semester st at the end of the 10th week of the semester
Two assignm	at the end of the 15th week of the semester nents each of <b>10 Marks</b> gnment at the end of 4th week of the semester
U	ssignment at the end of 9th week of the semester
Marks (dura	assion/Seminar/quiz any one of three suitably planned to attain the COs and POs for <b>2</b> <b>ration 01 hours</b> ) d of the 13th week of the semester
	three tests, two assignments, and quiz/seminar/group discussion will be out of 100 mark scaled down to 50 marks
(to have less methods of the CIE method the outcome	s stressed CIE, the portion of the syllabus should not be common /repeated for any of the cIE. Each method of CIE should have a different syllabus portion of the course). ds /question paper is designed to attain the different levels of Bloom's taxonomy as per e defined for the course. nd Examination:
Theory SEE papers for the	c will be conducted by University as per the scheduled timetable, with common question be subject ( <b>duration 03 hours</b> ) ion paper will have ten questions. Each question is set for 20 marks.
4 751	

# Suggested Learning Resources:

#### Text books: :

- 1. Introduction to Atomic Spectra", H.E. White, McGraw-Hill.
- 2. Fundamentals of Molecular Spectroscopy" C. N. Banwell, Tata McGraw-Hill
- 3. Atomic Physics", G. P. Harnwell & W.E. Stephens, McGraw-Hills Book Company, Inc.
- 4. Modern Spectroscopy", J. M. Hollas, John Wiley

#### **Reference books**:

- 1. Physics of Atoms and Molecules" by Bransden & Joachain, Pearson
- 2. Introduction to Spectroscopy" by Pavia et. al., Cengage Learning India Pvt. Ltd
- 3. Physics for degree students", C L Arora and P. S. Hemme, S Chand Publications

#### Web links and Video Lectures (e-Resources):

- 1. https://youtu.be/YrjJFQdzxfU
- 2. https://youtu.be/ds9EtslrEjM
- 3. https://youtu.be/0zDZqjqGtl4
- 4. https://youtu.be/Agu68RGaoWM

#### Activity Based Learning (Suggested Activities in Class)/ Practical Based learning

https://youtu.be/eZhpuzaIwNw

https://mas-iiith.vlabs.ac.in/List%20of%20experiments.html

**V** Semester

		GEO CHEMISTRY			
Course Co	de	21BSS532	CIE Marks	50	
Teaching Hours/Week (L:T:P: S)		2:1:0	SEE Marks	50	
Total Hour	rs of Pedagogy	25	Total Marks	100	
Credits		02	Exam Hours	03	
CLO 1	To introduce students t	o basic concepts of geocher	nistry and several up-	to-date issues	
	which are widely discus	sed in the field of geochemis	try.		
CLO 2	Explain the principles of	f optical mineralogy.			
CLO 3	Understand the elements	s of crystal chemistry and asp	pects of crystal structure	es	
CLO 4	Evaluate the formation of	of clay minerals, their classifi	cation, types & compos	sition	
CLO 5		•			
CLO 5       Understand the basic geochemistry of solid Earth.         Pedagogy (General Instructions)         These are sample Strategies, which teacher can use to accelerate the attainment of the various course outcomes.         These are sample Strategies, which teacher can use to accelerate the attainment of the various course outcomes.         1. Lecturer method (L) does not mean only traditional lecture method, but different type of teaching methods may be adopted to develop the outcomes.         2. Show Video/animation films to convince abstract concepts.         4. Encourage collaborative (Group Learning) Learning in the class         5. Ask at least three HOTS (Higher order Thinking) questions in the class, which promotes critical thinking         6. Adopt Problem Based Learning (PBL), which fosters students' Analytical skills, develop thinking skills such as the ability to evaluate, generalize, and analyse information rather than simply recall it.         7. Topics will be introduced in a multiple representation.         8. Show the different ways to solve the same problem and encourage the students to come up with their own creative ways to solve them.         9. Discuss how every concept can be applied to the real world - and when that's possible, it helps improve the students' understanding.         Module-1 Elements of Geochemistry         Omposition of elements, Layered structure of Earth and geochemistry Composition of different Earth reservoirs and the nuclides and radioactivity, Conservation of mass, isotopic and					
Pedagogy	Chalk and talk/pow	radiogenic isotopes in geoch verpoint presentation: Con ments, Concept of radioge	cepts of geochemistry	: Geochemical	
	<ul> <li>Videos/Learning material: Layered structure of Earth and geochemistry Composition of different Earth reservoirs and the nuclides and radioactivity</li> <li>Self-study: Conservation of mass, isotopic and elemental fractionation;</li> </ul>				
	v	· 1	,		
Minerals		<b>2 Rock forming minerals an</b> tion, physical and chemical	-	lassification of	
minerals. Compositie structures.	Composition of comm on of common rock-form	on oxides, carbonates, su ing minerals – Silicate and poscopy, Nature of light and p	lphides and sulphates non-silicate structures;	s, phosphates. CCP and HCP	
ropentes				ciul05y.	
PedagogyChalk and talk/power point presentation: Minerals - definition and classificatio physical and chemical properties. Chemical classification of minerals. Composition of common oxides, carbonates, sulphides and sulphates, phosphates. Videos/Learning material: Composition of common rock-forming minerals Silicate and non-silicate structures; CCP and HCP structures.			s. Composition		

<b>Self-study:</b> Properties of light and optical microscopy, Nature of light and				
	of optical mineralogy.			
Module-3 Crystallography				
indices, groups, I	rry ideas about crystal morphology in relation to internal structures, Crystal parameters and Crystal symmetry, projections and classification of crystals into six systems and 32 point Elements of crystal chemistry and aspects of crystal structures Stereographic projections of y elements and forms.			
Pedagogy	Chalk and talk/power point presentation: Elementary ideas about crystal morphology			
	in relation to internal structures, Crystal parameters and indices, Crystal symmetry,			
	projections, Elements of crystal chemistry and aspects of crystal structures			
	Videos/Learning material: Classification of crystals into six systems and 32 point			
	groups.			
	Self-study: Stereographic projections of symmetry elements and forms.			
	Module-4 Chemistry of rocks & minerals			
Organic	material in sediments; organic reactions, carbon in rocks, origin of petroleum, origin of coal,			
-	natter in black shale, carbon compounds as reducing agents.			
0	n of clay minerals, their classification, types, composition. Properties of soils.			
Pedagogy	Chalk and talk/power point presentation: Organic material in sediments; organic			
0.00	reactions, carbon in rocks, origin of petroleum, origin of coal, organic matter in black			
	shale, carbon compounds as reducing agents.			
Videos/Learning material: Formation of clay minerals, their classifie				
	composition. Properties of soils.			
Self-study: Carbon compounds as reducing agents.				
	Module-5 Geochemistry of Solid Earth			
the forma	d Earth – geochemical variability of magma and its products. The Earth in the solar system, ation of solar system. Composition of the bulk silicate Earth Meteorites. hical behavior of selected elements like Si, Al, K, Na etc			
	th as a physico-chemical system; Crust as a separate system, Geochemical cycle, entals of Radioactive and Radiogenic Isotope Geochemistry.			
Pedagogy	Chalk and talk/power point presentation: The solid Earth – geochemical variability of magma and its products. The Earth in the solar system, the formation of solar system. Composition of the bulk silicate Earth Meteorites.			
	Videos/Learning material: The earth as a physico-chemical system; Crust as a separate			
	system, Geochemical cycle, Fundamentals of Radioactive and Radiogenic Isotope			
	Geochemistry.			
	Self-study: Geochemical behavior of selected elements like Si, Al, K, Na etc.			
Course ou	tcome (Course Skill Set)			
	of the course the student will be able to:			
CO 1	Discuss the basics concepts of geochemistry			
CO 2	Understand the properties and identification of common rock-forming minerals.			
CO 3	Understand the crystal morphology, parameters & indices.			
<b>CO 4</b>	Enumerate the organic reactions and chemistry of rocks and minerals			
CO 5	Illustrate the geochemical behaviour & physico-chemical system of earth.			

The weightage of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is 50%. The minimum passing mark for the CIE is 40% of the maximum marks (20 marks). A student shall be deemed to have satisfied the academic requirements and earned the credits allotted to each subject/ course if the student secures not less than 35% (18 Marks out of 50) in the semester-end examination (SEE).

#### **Continuous Internal Evaluation:**

Three Unit Tests each of 20 Marks (duration 01 hour)

1. First test at the end of 5<sup>th</sup>week of the semester

2. Second test at the end of the  $10^{th}$  week of the semester

3. Third test at the end of the 15<sup>th</sup> week of the semester

#### Two assignments each of 10 Marks

4.First assignment at the end of 4th week of the semester

5. Second assignment at the end of 9th week of the semester Group discussion/Seminar/quiz any one of three suitably planned to attain the COs and POs for **20 Marks (duration 01 hours)** 

6. At the end of the 13<sup>th</sup>week of the semester The sum of three tests, two assignments, and quiz/seminar/group discussion will be out of 100 marks and will be scaled down to 50 marks (to have less stressed CIE, the portion of the syllabus should not be common /repeated for any of the methods of the CIE. Each method of CIE should have a different syllabus portion of the course).

CIE methods /question paper is designed to attain the different levels of Bloom's taxonomy as per the outcome defined for the course.

#### Semester End Examination:

Theory SEE will be conducted by University as per the scheduled timetable, with common question papers for the subject (duration 03 hours)

1. The question paper will have ten questions. Each question is set for 20 marks.

2. There will be **2 questions from each module**. Each of the two questions under a module (with a maximum of 3 subquestions), should have a mix of topics under that module. The students have to answer 5 full questions, selecting one full question from each module.

# Suggested Learning Resources:

Books

1. Kerr, P. F. (1959). Optical Mineralogy. McGraw-Hill. 3. Verma, P. K. (2010). Optical Mineralogy (Four Colour). Ane Books Pvt Ltd

2. Deer, W. A., Howie, R. A., & Zussman, J. (1992). An introduction to the rock-forming minerals (Vol. 696). London: Longman.

3. Walther, J. V. (2009). Essentials of geochemistry. Jones & Bartlett Publishers

4. Mason, B. (1986) Principles of Geochemistry. 3rd Edition, Wiley New York.

5. Faure, Gunter and Teresa M. Mensing (2004). Isotopes: Principles and Applications, Wiley India Pvt. Ltd.

#### Web links and Video Lectures (e-Resources):

- 1. https://www.youtube.com/watch?v=Xt9-iZw5l7w
- 2. https://www.youtube.com/watch?v=Zl8\_ya-8198

3. https://www.youtube.com/watch?v=fiMemypKqEI&list=PLHyuArGIIyyR\_2mObwQ3yng18LDnDqidp

#### Activity Based Learning (Suggested Activities in Class)/ Practical Based learning

• <u>https://mg-nitk.vlabs.ac.in/List%20of%20experiments.html</u>

#### **B.Sc. Honors (Physics/Chemistry)**

#### Choice Based Credit System (CBCS) and Outcome Based Education (OBE)

SEMESTER - V				
Infinite Series, Fourier analysis and Transforms.				
Course Code	21BSS533	CIE	50	
Teaching Hours/Week (L: T: P: S)	2:1:0:0	SEE	50	
Total Hours of Pedagogy	25	Total	100	
Credits	2	Exam Hours	3hrs	

# CEMESTED V

# **Course Learning Objectives:**

The course will enable students to:

- 1. Provide basic concepts of sequences and infinite series
- 2. Understand the basic concepts of Laplace transform and Fourier transform
- 3. Expand the periodic functions in terms of Fourier series
- 4. Solve the differential equations by using transform techniques

# **Pedagogy (General Instructions)**

These are sample Strategies; which teachers can use to accelerate the attainment of the various course outcomes.

- 1. In addition to the traditional lecture method, different types of innovative teaching methods may be adopted so that the delivered lessons shall develop students' theoretical and applied mathematical skills.
- 2. State the need for Mathematical Science Studies and Provide real-life examples.
- 3. Support and guide the students for self-study.
- 4. You will also be responsible for assigning homework, grading assignments and quizzes, and documenting students' progress.
- 5. Encourage the students for group learning to improve their creative and analytical skills.
- Show short related video lectures in the following ways 6.
  - As an introduction to new topics (pre-lecture activity).
  - As a revision of topics (post-lecture activity).
  - As additional examples (post-lecture activity).
  - As an additional material of challenging topics (pre-and post-lecture activity).
  - As a model solution of some exercises (post-lecture activity)

# **Module-1:** Infinite Series

Convergence and divergence of infinite series of positive real numbers, Necessary condition for convergence, Cauchy criterion for convergence; Tests for convergence of positive term series; Basic comparison test, Limit comparison test, D'Alembert's ratio test.

Self-Study: Cauchy's *n*<sup>th</sup>root.

# (RBT Levels: L1, L2 and L3)

5 hours

Chalk and talk method/PowerPoint Presentation. Pedagogy

# **Module-2: Laplace Transforms**

Definition of Laplace transform, Linearity, Laplace transforms of basic functions, derivatives and integrals, Shifting theorems, Change of scale property, Laplace transforms of periodic functions. Self-Study: Unit step function.

(RBT Level	s: L1, L2 and L3)	5 hours
Pedagogy	Chalk and talk method/PowerPoint Presentation.	
	Module-3: Fourier Series	
Periodic fu	nction, Dirichlet conditions, Fourier series, Half Range Fourier cosin	e and sine series,
Harmonic a		
•	: The complex form of Fourier series.	
(RBT Level	s: L1, L2 and L3)	5 hours
Pedagogy	Chalk and talk method/PowerPoint Presentation.	
	Module-4: Fourier Transforms	
Fourier and	l inverse Fourier transforms, Fourier sine and cosine transforms, Inverse	se Fourier sine and
cosine tran	sforms, Linearity property, Change of scale property, Shifting property	·.
Self-Study	: Modulation theorem.	
(RBT Level	s: L1, L2 and L3)	5 hours
Pedagogy	Chalk and talk method/PowerPoint Presentation.	
	Module-5: Solution of Equations by Fourier Transforms	
Solution of	f the integral equation by Fourier sine and cosine transforms, Convo	lution theorem for
Fourier tran	nsform, Fourier transform of Partial derivatives, Applications of infinite	Fourier transforms
	y value problems.	
•	: Finite Fourier transform.	
(RBT Level	s: L1, L2 and L3)	5 hours
Pedagogy	Chalk and talk method/PowerPoint Presentation.	
Course outo	zome:	
At the end of	f the course, the student will be able to:	
	erstand the nature of the series by various tests of an infinite series.	
1	ace transforms and its properties.	
	otain Fourier series expansion of periodic functions. ansform any function into algebraic functions using Fourier transform.	

• Solve the boundary value problems by Fourier transform techniques.

The weightage of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is 50%. The minimum passing mark for the CIE is 40% of the maximum marks (20 marks). A student shall be deemed to have satisfied the academic requirements and earned the credits allotted to each subject/ course if the student secures not less than 35% (18 Marks out of 50) in the semester-end examination(SEE).

# **Continuous Internal Evaluation:**

# Three Unit Tests each of 20 Marks (duration 01 hour)

- 1. First test at the end of 5th week of the semester
- 2. Second test at the end of the 10th week of the semester
- 3. Third test at the end of the 15th week of the semester

#### Two assignments each of 10 Marks

4. First assignment at the end of 4th week of the semester

5. Second assignment at the end of 9th week of the semester

Group discussion/Seminar/quiz any one of three suitably planned to attain the COs and POs for **20 Marks** (duration 01 hours)

6. At the end of the 13th week of the semester

The sum of three tests, two assignments, and quiz/seminar/group discussion will be out of 100 marks and will be scaled down to 50 marks

(to have less stressed CIE, the portion of the syllabus should not be common /repeated for any of the methods of the CIE. Each method of CIE should have a different syllabus portion of the course).

# CIE methods /question paper is designed to attain the different levels of Bloom's taxonomy as per the outcome defined for the course.

#### **Semester End Examination:**

Theory SEE will be conducted by University as per the scheduled timetable, with common question papers for the subject (**duration 03 hours**)

- 1. The question paper will have ten questions. Each question is set for 20 marks.
- 2. There will be 2 questions from each module. Each of the two questions under a module (with a maximum of 3 sub-questions), **should have a mix of topics** under that module.
- 3. The students have to answer 5 full questions, selecting one full question from each module.

#### **Books Recommended:**

- 1. James Ward Brown & Ruel V. Churchill (2011). *Fourier Series and Boundary Value Problems*. McGraw-Hill Education.
- 2. Robert G. Bartle & Donald R. Sherbert (2015). Introduction to Real Analysis (4th edition). Wiley India.
- 3. Charles K. Chui (1992). An Introduction to Wavelets. Academic Press.
- 4. Erwin Kreyszig (2011). Advanced Engineering Mathematics (10th edition). Wiley.
- 5. A. Zygmund (2002). Trigonometric Series (3rd edition). Cambridge University Press.
- 6. J. K. Goyal, K. P. Gupta, Gauri Shankar Gupta (2007). *Integral transforms*, 21st Edition, Pragati Prakashan.
- 7. B S Grewal (2021). Higher Engineering Mathematics, 44<sup>th</sup> Edition, Khanna Publishers.

# Web links and Video Lectures (e-Resources):

- <u>http://ocw.mit.edu/courses/mathematics/</u>
- <u>http://www.foureir-series.com/</u>
- <u>http://mathworld.wolfram.com/</u>

#### Activity Based Learning (Suggested Activities in Class)/ Practical Based Learning

- Quiz
- Group assignment
- Seminars

#### **V- Semester**

	Atomic & Mo	lecular and Semiconductor	r Physics Lab	
Course	Code	21BSPL54	CIE Marks	50
Teachi	ng Hours/Week (L:T:P:S)	1:0:2:0	SEE Marks	50
Credits	3	2	Exam Hours	3 Hours
	<ol> <li><b>a objectives:</b></li> <li>To learn the basics characteristic</li> <li>To study the spectral characteris</li> </ol>	1	6 1	š.
	<u>Experiments:</u> ght Experiments to be performed			
SI.NO		Experiments		
1	Energy gap of a semiconductor.			
2	Spectral response of photo diode its V	✓ –I characteristics		
3	Resistance measurement of a semicor	nductor by Vandes Pau's meth-	od	
4	Temperature coefficient of resistance	and energy gap of thermistor		
5	Characteristics of Zener diode & Zen	ner diode as voltage regulator		
6	Characteristics of FET			
7	Characteristics of MOSFET			
8	Transistor CE amplifier- study of frequency response and measurement gain			
9	Characteristics of transistor in CE configuration			
	Rydberg Constant:Wavelength of s	pectral lines of hydrogen		
10	Verification of Beer's law			
10 11	Verification of Beer's law			

#### Course outcomes (Course Skill Set):

At the end of the course the student will be able to:

- 1. To understand operation of semiconductor devices
- 2. Acquire the knowledge of various characteristics of electronic devices
- 3. Analyse the spectral series of Hydrogen
- 4. Verify and analyse Beer's law.

Assessment Details (both CIE and SEE)

Continuous Internal Evaluation (CIE): The CIE marks awarded in case of Practical shall be based on the weekly

evaluation of laboratory journals/ reports after the conduction of every experiment and one practical test.

Semester End Evaluation (SEE students): The practical examinations to be conducted as per the time table of

University in a batch wise with strength of students not more than 10-15 per batch.

- 1. All laboratory experiments are to be included for practical examination.
- 2. Breakup of marks and the instructions printed on the cover page of answer script to be strictly adhered by the examiners.
- 3. Students can pick one experiment from the questions lot prepared by the examiners.
- 4. Change of experiment is allowed only once and 15% Marks allotted to the procedure part to be made zero.

#### Books:

- 1. Advanced Practical Physics for students, B.L. Flint and H.T. Worsnop, 1971, Asia Publishing House
- 2. A Text Book of Practical Physics, I.Prakash and Ramakrishna, 11th Ed., 2011, Kitab Mahal
- 3. Advanced level Physics Practicals, Michael Nelson and Jon M. Ogborn, 4th Edition, reprinted 1985, Heinemann Educational Publishers
- 4. A Laboratory Manual of Physics for undergraduate classes, D.P.Khandelwal, 1985, Vani Pub
- 5. Geeta Sanon, BSc Practical Physics, 1st Edn. (2007), R. Chand & Co.
- 6. Nelson and Jon Ogborn, Practical Physics
- 7. Practical Physics, Gupta &kumar voli &I
- 8. B.Sc. Practical Physics, C. L. Arora, S. Chand company

Suggested Learning Resources: https://virtuallabs.merlot.org/vl\_physics.html https://www.myphysicslab.com https://nptel.ac.in/courses/115105121

# **B.Sc. Honors (Physics/Chemistry/Mathematics)**

# Choice Based Credit System (CBCS) and Outcome Based Education (OBE)

SEMESTER - V				
Subject Name: Concepts of CyberSecurity				
Course Code	21BSO551	CIE	50	
Teaching Hours/Week (L: T: P: S)	2:1:0:0	SEE	50	
Total Hours of Pedagogy	25	Total	100	
Credits	2	Exam Hours	3hrs	

# CEMECTED V

# **Course Learning Objectives:**

The course will enable students to:

- Learn the foundations of Cyber security and threat landscape.
- To equip students with the technical knowledge and skills needed to protect and defend against • cyber threats.
- To develop skills in students that can help them plan, implement, and monitor cyber security mechanisms to ensure the protection of information technology assets.
- To expose students to governance, regulatory, legal, economic, environmental, social and ethical contexts of cyber security.
- To expose students to responsible use of online social media networks.
- To systematically educate the necessity to understand the impact of cyber crimes and threats with solutions in a global and societal context.
- To select suitable ethical principles and commit to professional responsibilities and human values and contribute value and wealth for the benefit of the society

# **Pedagogy (General Instructions)**

These are sample Strategies; which teachers can use to accelerate the attainment of the various course outcomes.

- 1. In addition to the traditional lecture method, different types of innovative teaching methods may be adopted so that the delivered lessons shall develop students' theoretical and applied mathematical skills.
- 2. State the need for Mathematical Science Studies and Provide real-life examples.
- 3. Support and guide the students for self-study.
- 4. You will also be responsible for assigning homework, grading assignments and quizzes, and documenting students' progress.
- 5. Encourage the students for group learning to improve their creative and analytical skills.
- 6. Show short related video lectures in the following ways
  - As an introduction to new topics (pre-lecture activity).
  - As a revision of topics (post-lecture activity).
  - As additional examples (post-lecture activity).
  - As an additional material of challenging topics (pre-and post-lecture activity).
  - As a model solution of some exercises (post-lecture activity)

# **Module-1: Introduction to Cyber Security**

Defining Cyberspace and Overview of Computer and Web-technology, Architecture of cyberspace, Communication and web technology, Internet, World wide web, Advent of Internet, Internet infrastructure for data transfer and governance, Internet Society, Regulation of cyberspace, Concept of cyber security, Issues and challenges of cyber security.

# (RBT Levels: L1, L2)

Chalk and talk method/PowerPoint Presentation. Pedagogy

5 hours

2

# **Module-2: Cybercrime and Cyberlaw**

Classification of cyber crimes, Common Cyber crimes- cyber crime targeting computers and mobiles, cyber crime against women and children, financial frauds, social engineering attacks, malware and ransomware attacks, zero-day and zero click attacks, Cybercriminals modus-operandi, Reporting of cyber crimes, Remedial and mitigation measures, Legal perspective of cybercrime, IT Act 2000 and its amendments.

# (RBT Levels: L1, L2)

Pedagogy Chalk and talk method/PowerPoint Presentation.

# Module-3: Social Media Overview and Security

Introduction to Social Networks. Types of Social media monitoring, hashtags, Viral content, Social media marketing, Social media privacy, Challenges, opportunities and pitfalls in online social networks, Security reporting of inappropriate content.

(RBT Levels: L1, L2)

Chalk and talk method/PowerPoint Presentation. Pedagogy

# Module-4: E-Commerce and Digital Payments 1

Definition of E-commerce, Main components of E-Commerce threats, E-Commerce security best practices, Introduction to digital payments, Components of digital payment and stakeholders, Modes of digital payments- Banking Cards, Unified Payment Interface (UPI), e-Wallets, Unstructured Supplementary Service Data (USSD) and Aadhar enabled payments.

# (RBT Levels: L1, L2)

Pedagogy Chalk and talk method/PowerPoint Presentation.

# Module-5: Digital Payments-Frauds and Customer Protection

Digital payments related common frauds and preventive measures. RBI guidelines on digital payments and customer protection in unauthorised banking transactions, Relevant provisions of Payment Settlement Act, 2007.

# (RBT Levels: L1, L2 and L3)

Chalk and talk method/PowerPoint Presentation. Pedagogy

# **Course outcome:**

At the end of the course, the student will be able to:

- Understand the concept of Cyber security and the issues and challenges associated with it
- Understand the cyber crimes, their nature, legal remedies and as to how to report the crimes through available platforms and procedures.
- Appreciate various privacy and security concerns on online social media and understand the reporting procedure of inappropriate content, underlying legal aspects.
- understand the basic concepts related to E-Commerce and digital payments.
- They will become familiar with various digital payment modes and related cyber security aspects, RBI guidelines and preventive measures against digital payment frauds.

5 hours

5 hours

5 hours

5 hours

The weightage of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is 50%. The minimum passing mark for the CIE is 40% of the maximum marks (20 marks). A student shall be deemed to have satisfied the academic requirements and earned the credits allotted to each subject/ course if the student secures not less than 35% (18 Marks out of 50) in the semester-end examination(SEE).

### **Continuous Internal Evaluation:**

#### Three Unit Tests each of 20 Marks (duration 01 hour)

- 1. First test at the end of 5th week of the semester
- 2. Second test at the end of the 10th week of the semester

3. Third test at the end of the 15th week of the semester

#### Two assignments each of 10 Marks

4. First assignment at the end of 4th week of the semester

5. Second assignment at the end of 9th week of the semester

Group discussion/Seminar/quiz any one of three suitably planned to attain the COs and POs for **20 Marks** (duration 01 hour)

6. At the end of the 13th week of the semester

The sum of three tests, two assignments, and quiz/seminar/group discussion will be out of 100 marks and will be scaled down to 50 marks

(to have less stressed CIE, the portion of the syllabus should not be common /repeated for any of the methods of the CIE. Each method of CIE should have a different syllabus portion of the course).

# CIE methods /question paper is designed to attain the different levels of Bloom's taxonomy as per the outcome defined for the course.

#### **Semester-End Examination:**

Theory SEE will be conducted by University as per the scheduled timetable, with common question papers for the subject (**duration 03 hours**)

- 1. The question paper will have ten questions. Each question is set for 20 marks.
- 2. There will be 2 questions from each module. Each of the two questions under a module (with a maximum of 3 sub-questions), **should have a mix of topics** under that module.
- 3. The students have to answer 5 full questions, selecting one full question from each module.

#### **Books Recommended:**

- 1. Cyber Crime Impact in the New Millennium, by R C Mishra, Auther Press. Edition 2010.
- 2. Cyber Security Understanding Cyber Crimes, Computer Forensics and Legal Perspectives by Sumit Belapure and Nina Godbole, Wiley India Pvt. Ltd. (First Edition, 2011)
- 3. Security in the Digital Age: Social Media Security Threats and Vulnerabilities by Henry A Oliver, Create Space Independent Publishing Platform. (person, 13<sup>th</sup> November, 2001)
- 4. Electronic Commerce by Elias M Awad, Prentice Hall of India Pvt Ltd.
- 5. Cyber Laws: Intellectual Property & E-Commerce Security by Kumar K, Dominant Publishers.
- 6. Network security Bible, Eric Cole, Ronald Krutz, James W Conley, 2<sup>nd</sup> Edition, Wiley India Pvt. Ltd.
- 7. Fundamentals of Network Security by E Maiwald, McGraw hill.

Web links and Video Lectures (e-Resources):

- <u>https://www.bing.com/videos/riverview/relatedvideo?&q=introductio9n+to+cyber+security+videos&&mid=9D4475C0E2A498B7D7E09D4475C0E2A498B7D7E0&&FORM=VRDGAR</u>
- <u>https://www.bing.com/videos/riverview/relatedvideo?&q=Architecture+of+cyber+space&&mi</u> d=59423C509BEF87F66C5659423C509BEF87F66C56&&FORM=VRDGAR\
- <u>https://www.bing.com/videos/riverview/relatedvideo?&q=classifications+of+cyber+crimes&&</u> <u>mid=A90594668BF86AF7F414A90594668BF86AF7F414&&FORM=VRDGAR</u>
- <u>https://www.bing.com/videos/riverview/relatedvideo?&q=Introduction+to+social+networks&&</u> <u>mid=A253D1EE3D51683C6D28A253D1EE3D51683C6D28&&FORM=VRDGAR</u>

Activity-Based Learning (Suggested Activities in Class)/Practical-Based Learning

- Quiz
- Group assignment
- Seminars

# B.Sc. Honors (Physics/Chemistry/Mathematics) Choice Based Credit System (CBCS) and Outcome Based Education (OBE)

SEMESTER - V				
Subject Name: Data Science				
Course Code	21BSO552	CIE	50	
Teaching Hours/Week (L: T: P: S)	2:1:0:0	SEE	50	
Total Hours of Pedagogy	25	Total	100	
Credits	2	Exam Hours	3hrs	

# **Course Learning Objectives:**

The course will enable students to:

- 1. Understand the knowledge of Mathematics to explain the concept of data science
- 2. Design Decision tree to predict the class for a given data
- 3. Analyze the given data set, and solve a problem by performing Classifications using the basics of mathematics and the data science
- 4. Develop solutions to group entities in the data set and apply it to the given real-world data using the basic knowledge.

# **Pedagogy (General Instructions)**

These are sample Strategies; which teachers can use to accelerate the attainment of the various course outcomes.

- 1. In addition to the traditional lecture method, different types of innovative teaching methods may be adopted so that the delivered lessons shall develop students' theoretical and applied mathematical skills.
- 2. State the need for Mathematical Science Studies and Provide real-life examples.
- 3. Support and guide the students for self-study.
- 4. You will also be responsible for assigning homework, grading assignments and quizzes, and documenting students' progress.
- 5. Encourage the students for group learning to improve their creative and analytical skills.
- 6. Show short related video lectures in the following ways
  - As an introduction to new topics (pre-lecture activity).
  - As a revision of topics (post-lecture activity).
  - As additional examples (post-lecture activity).
  - As an additional material of challenging topics (pre-and post-lecture activity).
  - As a model solution of some exercises (post-lecture activity)

# **Module-1: Introduction**

Data-Analytic Thinking: The Ubiquity of Data Opportunities, Example: Hurricane Frances, Example: Predicting Customer Churn. Data Science, Engineering, and Data-Driven Decision Making, Data Processing and Big Data. Data and Data Science Capability as a Strategic Asset, Data-Analytic Thinking. **Text Book 1: Chapter 1** 

# (RBT Levels: L1, L2)

5 hours

Pedagogy Chalk and talk method/PowerPoint Presentation.

# Module-2: Business Problems and Data Science Solutions

From business problems to data mining tasks, supervised versus unsupervised methods, data mining and its results, the data mining process, business understanding, data understanding, data preparation, modelling, evaluation, deployment, and other analytics techniques and technologies: Statistics, database querying, data warehousing, regression analysis, machine learning and data mining. **Text Book 1: Chapter 2** 

(RBT Levels	s: L1, L2)	5 hours
Pedagogy Chalk and talk method/PowerPoint Presentation.		
	Module-3: Introduction to Predictive Mo	odeling
segmentation supervised	elation to supervised segmentation models, induction, selecting informative attributes example: Attribute s segmentation with Tree-structured models, visualizing ability estimation, Example: Addressing the churn Problem <b>Chapter 2</b>	selection with information gain, segmentations, Trees as sets of
(RBT Levels	s: L1, L2)	5 hours
Pedagogy	Chalk and talk method/PowerPoint Presentation.	
	Module-4:Fitting of a Model to Dat	ta
function, and scoring and probability	on via Mathematical functions: Linear Discriminant function example of mining a linear discriminant from data, I ranking instances, support machines briefly, regression estimation and logistic regression. Logistic regression: a gression versus Tree. Induction, non-linear functions, super <b>Chapter 4</b>	linear discriminant functions for via mathematical functions, class some technical details. Example:
(RBT Levels	s: L1, L2 and L3)	5 hours
Pedagogy	Chalk and talk method/PowerPoint Presentation.	
	Module-5: Overfitting and its Avoidar	nce:
From holdou	l concepts, exemplary techniques, regularization, overf at evaluation to cross-validation, the Churn dataset revis and complexity control. (chapter 5)	
(RBT Levels	s: L1, L2 and L3)	5 hours
Pedagogy	Chalk and talk method/PowerPoint Presentation.	
<ul> <li>Apply techn</li> <li>Deve data s</li> <li>Analy of ma</li> <li>Deve the ba</li> </ul>	ome: The course, the student will be able to: y the knowledge of mathematics to explain the concept of iques in data science and its scope in business elop a decision tree based on supervised segmentation and set by selecting the attribute for segmentation using the a tyze the given data set, and solve a problem by performin thematics and data science lop solutions to group entities in data set and apply it for asic knowledge. rstand the concepts of overfitting and curves.	nd predict the class for a given available techniques og classification using the basics

The weightage of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is 50%. The minimum passing mark for the CIE is 40% of the maximum marks (20 marks). A student shall be deemed to have satisfied the academic requirements and earned the credits allotted to each subject/ course if the student secures not less than 35% (18 Marks out of 50) in the semester-end examination(SEE).

# **Continuous Internal Evaluation:**

# Three Unit Tests each of 20 Marks (duration 01 hour)

- 1. First test at the end of 5th week of the semester
- 2. Second test at the end of the 10th week of the semester

3. Third test at the end of the 15th week of the semester

#### Two assignments each of 10 Marks

4. First assignment at the end of 4th week of the semester

5. Second assignment at the end of 9th week of the semester

Group discussion/Seminar/quiz any one of three suitably planned to attain the COs and POs for **20 Marks** (duration **01 hour**)

6. At the end of the 13th week of the semester

The sum of three tests, two assignments, and quiz/seminar/group discussion will be out of 100 marks and will be scaled down to 50 marks

(to have less stressed CIE, the portion of the syllabus should not be common /repeated for any of the methods of the CIE. Each method of CIE should have a different syllabus portion of the course).

# CIE methods /question paper is designed to attain the different levels of Bloom's taxonomy as per the outcome defined for the course.

#### **Semester-End Examination:**

Theory SEE will be conducted by University as per the scheduled timetable, with common question papers for the subject (**duration 03 hours**)

- 1. The question paper will have ten questions. Each question is set for 20 marks.
- 2. There will be 2 questions from each module. Each of the two questions under a module (with a maximum of 3 sub-questions), **should have a mix of topics** under that module.
- 3. The students have to answer 5 full questions, selecting one full question from each module.

# **Books Recommended:**

#### **Textbooks:**

1. Foster Provost and Tom Fawcett. Data Science for Business, O'Reilly, 2013. First edition.

# **Reference Books:**

- 1. Cathy O'Neil and Rachel Schutt, Doing Data Science, O'Reilly, 2013
- 2. Hector Cuesta, Practical Data Analysis, PACKT Publishing, 2013,
- 3. Michel R. Berthold, Christian Borgelt, Frank Hijppner Frank Klawonn, **Guide to Intelligent Data Analysis**, Springer-Verlag London Limited, 2010

# Web links and Video Lectures (e-Resources):

- <u>https://www.bing.com/videos/riverview/relatedvideo?&q=data+processing+and+bigdata+video</u> s&&mid=0739E0F892304324175F0739E0F892304324175F&&FORM=VRDGAR
- <u>https://www.bing.com/videos/riverview/relatedvideo?&q=Business+problems+and+data+scien</u> ce+solutions&&mid=0B32390DDED02C33EC900B32390DDED02C33EC90&&FORM=VR DGAR
- <u>https://www.bing.com/videos/riverview/relatedvideo?&q=Predective+modelling+videos&&mid=B7CCA25FFDEB60A52C04B7CCA25FFDEB60A52C04&&FORM=VRDGAR</u>

#### Activity-Based Learning (Suggested Activities in Class)/Practical-Based Learning

- Quiz
- Group assignment
- Seminars

# **B.Sc. Honors (Physics/Chemistry/Mathematics)**

# Choice Based Credit System (CBCS) and Outcome Based Education (OBE)

	SEMESTER - V		
Sub	ject Name: Food and Nutrition	1	
Course Code	21BSO553	CIE	50
Teaching Hours/Week (L: T: P: S)	2:1:0:0	SEE	50
Total Hours of Pedagogy	25	Total	100
Credits	2	Exam Hours	3hrs

# **Course Learning Objectives:**

The course will enable students to:

- 1. To familiarize students with fundamentals of food, nutrients and their relationship to Health
- 2. To create awareness with respect to deriving maximum benefit from available food resources
- 3. Obtain knowledge of different food groups and their nutritive value and role in day's diet.
- 4. Understand the principles underlying Food Preparation.
- 5. Develop skills and techniques in Food Preparation with conservation of nutrients and Palatability using cooking methods generally employed.

# **Pedagogy (General Instructions)**

These are sample Strategies; which teachers can use to accelerate the attainment of the various course outcomes.

- 1. In addition to the traditional lecture method, different types of innovative teaching methods may be adopted so that the delivered lessons shall develop students' theoretical and applied mathematical skills.
- 2. State the need for Mathematical Science Studies and Provide real-life examples.
- 3. Support and guide the students for self–study.
- 4. You will also be responsible for assigning homework, grading assignments and quizzes, and documenting students' progress.
- 5. Encourage the students for group learning to improve their creative and analytical skills.
- 6. Show short related video lectures in the following ways
  - As an introduction to new topics (pre-lecture activity).
  - As a revision of topics (post-lecture activity).
  - As additional examples (post-lecture activity).
  - As an additional material of challenging topics (pre-and post-lecture activity).
  - As a model solution of some exercises (post-lecture activity)

# Module-1: INTRODUCTION TO FOODS

Classification of Food group: Basic 4, 5 and 7 food groups; functional food groups-energy yielding, body building and protective foods (only sources and not properties and functions), food pyramid. Study of various cooking methods - Boiling, steaming, stewing, frying, baking, roasting, broiling, cooking under pressure.

# (RBT Levels: L1, L2)

5 hours

Pedagogy Chalk and talk method/PowerPoint Presentation.

# Module-2: CEREALS AND PULSES

**Cereals** - composition of rice, wheat, effects of cooking on parboiled and raw rice, principles of starch cookery, gelatinization.

**Pulses**-Varieties of pulses and grams, composition, nutritive value, cooking quality of pulses, germination and its effect.

(RBT Levels: I	L1, L2)	5 hours
Pedagogy	Chalk and talk method/PowerPoint Presentation	ND
Tedagogy	Module-3: VEGETABLES, FRUITS, MILK	
methods and p Fruits -Compo enzymatic brow Milk - Compo changes in mil Beverages - C	Classification, composition, nutritive value, selectrinciples involved in cooking. Distion, nutritive value, changes during ripening	ction and preparation for cooking, , methods and effects of cooking, tion and homogenization of milk, and milk powder. ges- methods of preparing tea and on–alcoholic beverages.
(RBT Levels: I	L1, L2)	5 hours
Pedagogy C	halk and talk method/PowerPoint Presentation.	
M	lodule-4: FATS AND OILS, EGG, MEAT AN	ND MEAT PRODUCTS
Egg - Structure, Meat and mea	Types of oils, function of fats and oils, shortenin, composition, selection, nutritive value, uses of <b>t products</b> -Structure, composition, nutritive v t, aging, tenderness. Fish - Structure, composition, <b>1</b> L2)	egg in cookery, methods of cooking. alue, selection of meat, post-mortem
•	halk and talk method/PowerPoint Presentation.	5 110015
redagogy C		
	Module-5: NUTRITION	Ň
energy value o	- General introduction, history of Nutrition. Ene of foods, determination, physiological fuel value tors influencing BMR. Recommended Dietary A nins.	s, SDA of foods, basal metabolic rate-
(RBT Levels: I		5 hours
Pedagogy Ch	alk and talk method/PowerPoint Presentation.	
<ul> <li>To gain methods</li> <li>To gain and puls</li> <li>To get c fruits an the comp</li> <li>To have of egg proof egg proof on methods</li> </ul>	e course, the student will be able to: knowledge on food groups and its function, food knowledge on nutritive value, understand the o	cookery concepts involved in cereals nderstand the changes in pigments of of beverages. To have an overview of e preparation of milk. e and develop skills in the preparation value, selection and apply knowledge s and abuses of spices and condiments

The weightage of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is 50%. The minimum passing mark for the CIE is 40% of the maximum marks (20 marks). A student shall be deemed to have satisfied the academic requirements and earned the credits allotted to each subject/ course if the student secures not less than 35% (18 Marks out of 50) in the semester-end examination(SEE).

# **Continuous Internal Evaluation:**

### Three Unit Tests each of 20 Marks (duration 01 hour)

- 1. First test at the end of 5th week of the semester
- 2. Second test at the end of the 10th week of the semester
- 3. Third test at the end of the 15th week of the semester

#### Two assignments each of 10 Marks

4. First assignment at the end of 4th week of the semester

5. Second assignment at the end of 9th week of the semester

Group discussion/Seminar/quiz any one of three suitably planned to attain the COs and POs for **20 Marks** (duration **01 hour**)

6. At the end of the 13th week of the semester

The sum of three tests, two assignments, and quiz/seminar/group discussion will be out of 100 marks and will be scaled down to 50 marks

(to have less stressed CIE, the portion of the syllabus should not be common /repeated for any of the methods of the CIE. Each method of CIE should have a different syllabus portion of the course).

# CIE methods /question paper is designed to attain the different levels of Bloom's taxonomy as per the outcome defined for the course.

#### **Semester-End Examination:**

Theory SEE will be conducted by University as per the scheduled timetable, with common question papers for the subject (**duration 03 hours**)

- 1. The question paper will have ten questions. Each question is set for 20 marks.
- 2. There will be 2 questions from each module. Each of the two questions under a module (with a maximum of 3 sub-questions), **should have a mix of topics** under that module.
- 3. The students have to answer 5 full questions, selecting one full question from each module.

# **Books Recommended:**

- 1. Srilakshmi, B., Food Science, (2016), 5<sup>th</sup> edition, New Age Publishers, India, New Delhi.
- 2. Srilakshmi, B. (2017) Nutrition Science, New Age International (P) Ltd., New Delhi.
- 3. Many, S and Shadaksharaswami, M. (2008) Food: Facts and Principles, 3rd edition, New Age Publishers
- 4. Swaminathan, M., (2012) Food Science, Chemistry and Experimental foods, Bangalore Printing and Publishing Company.
- 5. Potter M,N. and Hotchkiss, J.H. (1998) Food Science 5<sup>th</sup> edition, CBS Publications and Distributors, Daryaganji, New Delhi.
- 6. Philip, T., Modern Cookery for teaching and trade, volume I and II, Orient Longmans Ltd.

# Web links and Video Lectures (e-Resources):

- <u>www.nal.vsda.gov/fnic/foodcomp</u>
- <u>www.fda.gov-vegetables</u>
- <u>http://www.eatforhealth.gov.au-fleshfoods,egg&milk</u>
- https://www.business.qld.gov.av-sensoryanalysis of food products
- <u>https://youtu.be/oE8YV2zlO8M</u>

#### Activity-Based Learning (Suggested Activities in Class)/Practical-Based Learning

- Quiz
- Group assignment
- Seminars

# B.Sc. Honors (Physics/Chemistry/Mathematics)

# Choice Based Credit System (CBCS) and Outcome Based Education (OBE)

S	EMESTER - V		
Subje	ct Name: Indian Hist	ory	
Course Code	21BSO61	CIE	50
Teaching Hours/Week (L: T: P: S)	2:1:0:0	SEE	50
Total Hours of Pedagogy	25	Total	100
Credits	2	Exam Hours	3hrs
Course Learning Objectives:			
The course will enable students to:			
1. Learn about ancient India and its i	mportance in present a	allowance.	

- 2. Rulers of Ancient India and their contribution to Indian art and architecture in nativity.
- 3. Delhi Sultanate and their contributions, and religious moments of medieval India to enlighten Indian cultural importance.
- 4. Mughals Administration and their contribution to Indian architecture, Rise of Indian dynasty's against foreign invaders (Europeans).
- 5. Struggle to establish independent freedom India by our freedom fighters.

# **Pedagogy (General Instructions)**

These are sample Strategies; which teachers can use to accelerate the attainment of the various course outcomes.

- 1. In addition to the traditional lecture method, different types of innovative teaching methods may be adopted so that the delivered lessons shall develop students' theoretical and applied mathematical skills.
- 2. State the need for Mathematical Science Studies and Provide real-life examples.
- 3. Support and guide the students for self–study.
- 4. You will also be responsible for assigning homework, grading assignments and quizzes, and documenting students' progress.
- 5. Encourage the students for group learning to improve their creative and analytical skills.
- 6. Show short related video lectures in the following ways
  - As an introduction to new topics (pre-lecture activity).
  - As a revision of topics (post-lecture activity).
  - As additional examples (post-lecture activity).
  - As an additional material of challenging topics (pre-and post-lecture activity).
  - As a model solution of some exercises (post-lecture activity)

# **Module-1: Ancient India**

Ancient India : Sources of ancient Indian history, Indus Valley Civilization and culture, Vedic Civilization and culture, Religious movements of Ancient India (Jainism and Buddhism).

# (RBT Levels: L1, L2)

5 hours

Pedagogy Chalk and talk method/PowerPoint Presentation.

# Module-2: Rulers of Ancient India

Rulers of Ancient India : Mauryan Empire - Chandragupta Maurya and Ashoka, Rise of Gupta Empire - Samudragupta and Chandragupta II, Cultural developments in Gupta age, States of South India in brief, Rise of Rajput States - Prithviraj Chauhan, Maharana Pratap. Social and Cultural Development during Rajput Dynasty.

# (RBT Levels: L1, L2)

5 hours

Pedagogy

Chalk and talk method/PowerPoint Presentation.

# **Module-3: Medieval India**

Medieval India: Banavasi Kadambas - Mayura Varma, Badami Chalukyas - Pulakeshi II, The Rashtrakutas - Amoghavarsha & Hoysalas - Vishnuvardana, Alauddin Khilji, Art and architecture under Sultanate rule, Vijayanagara empire - Krishnadevaraya, Architecture and Literature during Vijayanagara Empire, Religious moments of medieval India (Bhakti saints and sufism) - Acharya's, Ramanand, Kabir, Mirabai, Chaitanya, Guru Nanak).

# (RBT Levels: L1, L2 and L3)

Chalk and talk method/PowerPoint Presentation. Pedagogy

# Module-4: Post Medieval India & Advent of the Europeans

Post Medieval India : Akbar, Shahjahan, Mughal architecture and culture, Nayakas of Keladi - Shivappa nayaka and Rani Chennamma, Mysore Wodeyars - Nalwadi Krishna Raja Wodeyar, Sir M Visvesvaraya, The Mughal and Maratha conflict - Chhatrapati Shivaji, the peshwa's, Advent of the Europeans, Rise of the British - Conquest of Bengal, Battle of Plassey, Battle of Buxar, Anglo Mysore Wars, Anglo Maratha Wars.

# (RBT Levels: L1, L2 and L3)

Chalk and talk method/PowerPoint Presentation. Pedagogy

# Module-5: The revolt of 1857 to Towards till 1947.

The revolt of 1857, Economic impact of British rule, Partition of Bengal and Swadeshi movement, Mahatma Gandhi, Civil disobedience movement, Quit India Movement, Impact of the Second World War - 1945-47, Towards freedom - August 15th 1947.

#### (RBT Levels: L1, L2 and L3)

Chalk and talk method/PowerPoint Presentation. Pedagogy

# **Course outcome:**

At the end of the course, the student will be able to analyse and understand about:

- Our Ancient India and its importance in present allowance.
- Kings of Ancient India and their contribution to Indian art and architecture.
- Delhi Sultanate and their contributions, religious movements of medieval India to enlighten Indian cultural importance.
- Mughals Administration and their contribution to Indian architecture, Rise of Indian dynasty's against foreign invaders (Europeans).
- Today's Independent freedom India by the efforts of our freedom fighters.

5 hours

5 hours

5 hours

The weightage of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is 50%. The minimum passing mark for the CIE is 40% of the maximum marks (20 marks). A student shall be deemed to have satisfied the academic requirements and earned the credits allotted to each subject/ course if the student secures not less than 35% (18 Marks out of 50) in the semester-end examination (SEE).

# **Continuous Internal Evaluation:**

### Three Unit Tests each of 20 Marks (duration 01 hour)

- 1. First test at the end of 5th week of the semester
- 2. Second test at the end of the 10th week of the semester
- 3. Third test at the end of the 15th week of the semester

#### Two assignments each of 10 Marks

4. First assignment at the end of 4th week of the semester

5. Second assignment at the end of 9th week of the semester

Group discussion/Seminar/quiz any one of three suitably planned to attain the COs and POs for **20 Marks** (duration **01 hour**)

6. At the end of the 13th week of the semester

The sum of three tests, two assignments, and quiz/seminar/group discussion will be out of 100 marks and will be scaled down to 50 marks

(to have less stressed CIE, the portion of the syllabus should not be common /repeated for any of the methods of the CIE. Each method of CIE should have a different syllabus portion of the course).

# CIE methods /question paper is designed to attain the different levels of Bloom's taxonomy as per the outcome defined for the course.

# **Semester-End Examination:**

Theory SEE will be conducted by University as per the scheduled timetable, with common question papers for the subject (**duration 03 hours**)

- 1. The question paper will have ten questions. Each question is set for 20 marks.
- 2. There will be 2 questions from each module. Each of the two questions under a module (with a maximum of 3 sub-questions), **should have a mix of topics** under that module.
- 3. The students have to answer 5 full questions, selecting one full question from each module.

# **Books Recommended:**

# **Text Books:**

- 1. India's Ancient Past, R S Sharma, Oxford University Press, INDIA.
- 2. A History of Medieval India by Satish Chandra, Orient BlackSwan Pvt. Ltd.; Revised edition (7 July 2020).
- 3. History of Modern India by Bipan Chandra, Orient BlackSwan Pvt. Ltd.; Revised edition (7 July 2020).

# **Reference Books:**

- 1. Prehistory and Protohistory of India An Appraisal by VK Jain, D.K. Print World Ltd; 1st edition (1 June 2006).
- 2. Ancient History of India by Charles J. Naegele
- 3. History of Medieval India: From 1000 A. D. To 1707 A. D. by R.S. Chaurasia
- 4. From Plassey To Partition: A History Of Modern India by Sekhar Bandyopadhyay.

#### Activity-Based Learning (Suggested Activities in Class)/Practical-Based Learning

- Quiz
- Group assignment
- Seminars

# **B.Sc. Honors (Physics/Chemistry/Mathematics)** Choice Based Credit System (CBCS) and Outcome Based Education (OBE)

	SEMESTER - V		
	Subject Name: Economics		
Course Code	21BSO562	CIE	50
Teaching Hours/Week (L: T: P: S)	2:1:0:0	SEE	50
Total Hours of Pedagogy	25	Total	100
Credits	2	Exam Hours	3hrs
Course Learning Objectives:			
The course will enable students to:	· · · · · · · · · · · · · · · · · · ·	1	
1. This course aims at providir		ed concepts of engined	ering economic
analysis and its role in engine	5	vistion often tax analys	ia ranlaaamant
2. Additionally, the course also		-	sis, replacement
analysis, uncertainty, inflatio Pedagogy (General Instructions)	ii, defiation, and estimation	of future events.	
These are sample Strategies; which t	eachers can use to accelerat	e the attainment of the	various course
outcomes.	eachers can use to accelerate	e the attainment of the	various course
1. In addition to the traditional	lecture method. different ty	pes of innovative teacl	hing methods
may be adopted so that the c	•	-	-
mathematical skills.		op students theoretice	and applied
	tical Caianaa Chudiaa and Du	avida nasl life avamul	• •
2. State the need for Mathema		ovide real-me example	es.
3. Support and guide the stude	•		
4. You will also be responsible		rading assignments an	d
quizzes, and documenting st	1 0		
5. Encourage the students for g	group learning to improve th	eir creative and analyt	ical skills.
6. Show short related video le	ectures in the following way	S	
• As an introduction to new	v topics (pre-lecture activity	).	
• As a revision of topics (p	ost-lecture activity).		
• As additional examples (	-		
	of challenging topics (pre-a	and post-lecture activit	v).
	me exercises (post-lecture a	-	5)*
Modu	ale-1: Introduction to Ecor	nomics	
Nature and Scope of Economics,	Basic Concepts in Econor	nics, Micro and Mac	cro Economics,
Importance of Study of Economic	-		
Economics			
(RBT Levels: L1, L2)			5 hours
· , ,	Down Doint Dugg		5 11001 5
8 05	/PowerPoint Presentation.	_	
	odule-2: Demand and Sup		
Demand – Types – Determinants – Supply – Market price determinatio Consumer Survey – Trend Projectio	n – Case Study in Demand		
(RBT Levels: L1, L2 and L3)			5 hours
, , , , , , , , , , , , , , , , , , , ,			

# Pedagogy Chalk and talk method/PowerPoint Presentation.

# Module-3: Cost, Revenue, Market Structure and Market Failure

**Cost and Revenue:** Concepts – Classifications – Short run and long-run cost curves – Revenue – Concepts – Measurement of Profit (Case Study).

**Market Structure:** Perfect Competition – Characteristics – Price and output determination in short run and long run – Monopoly – Price Discrimination – Monopolistic Competition – Product Differentiation – Oligopoly and Duopoly.

**Market Failure**: Causes – Type of Goods – Rivalrous and Non-rivalrous goods – Excludable and Non-excludable goods – Solutions – Government Intervention.

(RBT Levels: L1, L2 and L3)

5 hours

5 hours

Pedagogy Chalk and talk method/PowerPoint Presentation.

# Module-4: Money and Banking

Money – Functions – Quantity theory of money – Banking – Commercial Banks – Functions – Central Bank (RBI) – Functions – Role of Banks in Economic Development.

(RBT Levels: L1, L2 and L3)

Pedagogy

chalk and talk method/PowerPoint Presentation.

# Module-5: Business Cycle and National Income

**Business Cycle and National Income**: Meaning –Phases of business cycle - Inflation – Causes – Control measures - Deflation –National Income- Concepts – Methods of calculating national income – Problems in calculating national income.

(RBT Levels: L1, L2 and L3)

5 hours

Pedagogy Chalk and talk method/PowerPoint Presentation.

# Course outcome:

At the end of the course, the student will be able to:

- Describe the role of economics in the decision-making process and perform calculations in regard to interest formulas.
- Estimate the Present, annual and future worth comparisons for cash flows.
- Calculate the rate of return, depreciation charges and income taxes. Enumerate different cost entities in estimation and costing.
- Explain the importance of finance functions, financial ratios and solve related problems.
- Explain the elements of budgeting and benchmarking

The weightage of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is 50%. The minimum passing mark for the CIE is 40% of the maximum marks (20 marks). A student shall be deemed to have satisfied the academic requirements and earned the credits allotted to each subject/ course if the student secures not less than 35% (18 Marks out of 50) in the semester-end examination(SEE).

# **Continuous Internal Evaluation:**

### Three Unit Tests each of 20 Marks (duration 01 hour)

- 1. First test at the end of 5th week of the semester
- 2. Second test at the end of the 10th week of the semester

3. Third test at the end of the 15th week of the semester

#### Two assignments each of 10 Marks

4. First assignment at the end of 4th week of the semester

5. Second assignment at the end of 9th week of the semester

Group discussion/Seminar/quiz any one of three suitably planned to attain the COs and POs for **20 Marks** (duration 01 hour)

6. At the end of the 13th week of the semester

The sum of three tests, two assignments, and quiz/seminar/group discussion will be out of 100 marks and will be scaled down to 50 marks

(to have less stressed CIE, the portion of the syllabus should not be common /repeated for any of the methods of the CIE. Each method of CIE should have a different syllabus portion of the course).

# CIE methods /question paper is designed to attain the different levels of Bloom's taxonomy as per the outcome defined for the course.

#### **Semester-End Examination:**

Theory SEE will be conducted by University as per the scheduled timetable, with common question papers for the subject (**duration 03 hours**)

- 1. The question paper will have ten questions. Each question is set for 20 marks.
- 2. There will be 2 questions from each module. Each of the two questions under a module (with a maximum of 3 sub-questions), **should have a mix of topics** under that module.
- 3. The students have to answer 5 full questions, selecting one full question from each module.

# **Books Recommended:**

# **Text Books:**

- 1. Dewett. K.K., Navalur M. H., "Modern Economic Theory", S. Chand and Company Ltd, New Delhi, 24<sup>th</sup> Edn., 2014.
- 2. Lipsey & Chrystal, "Economics", Oxford University Press, 2010

# **Reference Books:**

- 1. Paul A Samuelson & William, "Economics", Tata McGraw Hill, New Delhi, 2012.
- 2. Francis Cherinullem "International Economics", McGraw Hill Education, 2011.
- 3. William A McEachern and Simrit Kaur, "Micro ECON", Cengage Learning, 2013.
- 4. William A McEachern and Indira A., "Macro ECON", Cengage Learning, 2014.

# Web links and Video Lectures (e-Resources):

# Activity-Based Learning (Suggested Activities in Class)/Practical-Based Learning

# • Quiz

- Group assignment
- Seminars

# **B.Sc. Honors (Physics/Chemistry/Mathematics)**

# Choice Based Credit System (CBCS) and Outcome Based Education (OBE)

	SEMESTER - V		
Subjec	t Name: Research Method	lology	
Course Code	21BRM57	CIE	50
Teaching Hours/Week (L: T: P: S)	2:1:0:0	SEE	50
Total Hours of Pedagogy	25	Total	100
Credits	2	Exam Hours	3hrs

# **Course Learning Objectives:**

The course will enable students to:

- Understand the knowledge on basics of research and its types.
- Understand the research design and its concepts.
- Understand methods of research analysis and report preparation.
- To Learn the concept of Literature Review, Pedagogy, Attributions and Citations and learn ethics in research.

# **Pedagogy (General Instructions)**

Teaching-Learning Process (General Instructions) These are sample Strategies; which teachers can use to accelerate the attainment of the various course outcomes.

- 1. Lecturer methods (L) need not be only the traditional lecture methods, but alternative effective teaching methods could be adopted to attain the outcomes.
- 2. Use of Video to explain various concepts on IPR.
- 3. Encourage collaborative (Group Learning) Learning in the class.
- 4. Ask at least three HOT (Higher Order Thinking) questions in the class, which promotes critical thinking.
- 5. Introduce Topics in manifold representations.
- 6. Show the different ways to analyze the research problem and encourage the students to come up with their own creative ways to solve them.
- 7. Discuss how every concept can be applied to the real world and when that's possible, it helps improve the student's understanding.

Module-1
Introduction: Foundations of Research: Meaning, Objectives, Motivation, Utility. Concept of
theory, empiricism, deductive and inductive theory. Characteristics of scientific method -
Understanding the language of research - Concept, Construct, Definition, Variable. Research
Process.
Problem Identification & Formulation – Research Question – Investigation Question – Measurement
Issues – Hypothesis – Qualities of a good Hypothesis –Null Hypothesis & Alternative Hypothesis.
Hypothesis Testing – Logic & Importance.
Teaching- Learning Process: Chalk and talk method / PowerPoint Presentation.
(RBT Levels: L1, L2) 5 hours
Module-2
Research Design: Concept and Importance in Research - Features of a good research design -
Exploratory Research Design - concept, types and uses, Descriptive Research Designs - concept, types
and uses. Experimental Design: Concept of Independent & Dependent variables.
Qualitative and Quantitative Research: Qualitative research - Quantitative research - Concept of
measurement, causality, generalization, replication. Merging the two approaches.
Teaching-Learning Process Chalk and talk method / Power Point Presentation
(RBT Levels: L1, L2) 5 hours

2

<b>Measurement</b> : Concept of measurement– what is measured? Problems in measurement in research –
1
Validity and Reliability. Levels of measurement – Nominal, Ordinal, Interval, Ratio.
Sampling: Concepts of Statistical Population, Sample, Sampling Frame, Sampling Error, Sample Size,
Non Response. Characteristics of a good sample. Probability Sample – Simple Random Sample,
Systematic Sample, Stratified Random Sample & Multi-stage sampling. Determining size of the sample
– Practical considerations in sampling and sample size.
Teaching-Learning Process : Chalk and talk method / PowerPoint Presentation
(RBT Levels: L1, L2) 5 hours
Module-4
Data Analysis: Data Preparation – Univariate analysis (frequency tables, bar charts, pie charts,
percentages), Bivariate analysis – Cross tabulations and Chi-square test including testing hypothesis of
association.
Interpretation of Data and Paper Writing – Layout of a Research Paper, Journals in Computer
Science, Impact factor of Journals, When and where to publish? Ethical issues related to publishing,
Plagiarism and Self-Plagiarism.
Teaching-Learning Process: Chalk and talk method/PowerPoint Presentation
(RBT Levels: L1, L2) 5 hours
(RBT Levels: L1, L2) 5 hours
(RBT Levels: L1, L2)     5 hours       Module-5:     5
(RBT Levels: L1, L2)5 hoursModule-5:Basic Principles of Design Rights - Use of Encyclopaedias', Research Guides, Handbook etc.,
(RBT Levels: L1, L2 )5 hoursModule-5:Basic Principles of Design Rights - Use of Encyclopaedias', Research Guides, Handbook etc., Academic Databases for Computer Science Discipline.
(RBT Levels: L1, L2 )5 hoursModule-5:Basic Principles of Design Rights - Use of Encyclopaedias', Research Guides, Handbook etc., Academic Databases for Computer Science Discipline.Use of tools/techniques for Research: methods to search required information effectively, Reference
(RBT Levels: L1, L2 )5 hoursModule-5:5Basic Principles of Design Rights - Use of Encyclopaedias', Research Guides, Handbook etc., Academic Databases for Computer Science Discipline.Use of tools/techniques for Research: methods to search required information effectively, Reference Management Software like Zotero/Mendeley, Software for paper formatting like LaTeX/MS Office,
(RBT Levels: L1, L2 )5 hoursModule-5:Module-5:Basic Principles of Design Rights - Use of Encyclopaedias', Research Guides, Handbook etc., Academic Databases for Computer Science Discipline.Use of tools/techniques for Research: methods to search required information effectively, Reference Management Software like Zotero/Mendeley, Software for paper formatting like LaTeX/MS Office, Software for detection of Plagiarism.
(RBT Levels: L1, L2 )5 hoursModule-5:Basic Principles of Design Rights - Use of Encyclopaedias', Research Guides, Handbook etc., Academic Databases for Computer Science Discipline.Use of tools/techniques for Research: methods to search required information effectively, Reference Management Software like Zotero/Mendeley, Software for paper formatting like LaTeX/MS Office, Software for detection of Plagiarism.Teaching-Learning Process: Chalk and talk method / PowerPoint Presentation (RBT Levels: L1, L2)5 hoursCourse outcome:5 hours
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(RBT Levels: L1, L2 )5 hoursModule-5:Basic Principles of Design Rights - Use of Encyclopaedias', Research Guides, Handbook etc., Academic Databases for Computer Science Discipline.Use of tools/techniques for Research: methods to search required information effectively, Reference Management Software like Zotero/Mendeley, Software for paper formatting like LaTeX/MS Office, Software for detection of Plagiarism.Teaching-Learning Process: Chalk and talk method / PowerPoint Presentation (RBT Levels: L1, L2)5 hoursCourse outcome:5 hours

CO 3. To know the concepts of measurement and saplings.

CO 4. To Understand the data analysis and interpretation.

CO 5. To Understand the tools and techniques for report preparation.

# Assessment Details (both CIE and SEE)

The weightage of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is 50%. The minimum passing mark for the CIE is 40% of the maximum marks (20 marks). A student shall be deemed to have satisfied the academic requirements and earned the credits allotted to each subject/ course if the student secures not less than 35% (18 Marks out of 50) in the semester-end examination (SEE).

# **Continuous Internal Evaluation:**

Three Unit Tests each of **20 Marks (duration 01 hour)** 

1. First test at the end of 5th week of the semester

2. Second test at the end of the 10th week of the semester

3. Third test at the end of the 15th week of the semester

Two assignments each of 10 Marks

4. First assignment at the end of 4th week of the semester

5. Second assignment at the end of 9th week of the semester
Group discussion/Seminar/quiz any one of three suitably planned to attain the COs and POs for 20 Marks (duration 01 hour)
6. At the end of the 13th week of the semester
The sum of three tests, two assignments, and quiz/seminar/group discussion will be out of 100 marks and will be scaled down to 50 marks
(to have less stressed CIE, the portion of the syllabus should not be common /repeated for any of the methods of the CIE. Each method of CIE should have a different syllabus portion of the course).
CIE methods /question paper is designed to attain the different levels of Bloom's taxonomy as per the outcome defined for the course.
Semester-End Examination:
Theory SEE will be conducted by University as per the scheduled timetable, with common question papers for the subject (duration 03 hours)

- 1. The question paper will have ten questions. Each question is set for 20 marks.
- 2. There will be 2 questions from each module. Each of the two questions under a module (with a maximum of 3 sub-questions), **should have a mix of topics** under that module.
- 3. The students have to answer 5 full questions, selecting one full question from each module.

#### **Books Recommended:**

Textbook 1. Dr Deepak Chawla & Dr Neena Sondhi "Research Methodology",

**ISBN:** 9789325982390, Vikas Publishing (2023)

Research Methodology - C.R.Kothari, Edition:2018

Research Methodology (Methods, Approaches and Techniques) by Dr. Baidyanath Mishra

published by Choukambha Orientalia, Edition:2018, ISBN:9788176373896

### **References:**

David V. Thiel "Research Methods for Engineers" Cambridge University Press, 978-1-107-03488- 4 - Activity Based Learning

# Web links and Video Lectures (e-Resources):

https://www.google.com/search?rlz=1C1ASVC\_enIN953IN954&q=%22weblinks%22+for+Research+method ology+and+IPR&sa=X&ved=2ahUKEwirt8XRhZiAAxVQb2wGHW9SB6QQ5t4CegQIOhAB https://www.dolphininstitute.in/workshops-seminars-conducted-on-research-methodology-ipr-andentrepreneurship/

http://www.cs.princeton.edu/courses/archive/fall02/cs526/

#### Activity-Based Learning (Suggested Activities in Class)/Practical-Based Learning

- Quiz
- Group assignment
- Seminars