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Third Semester B.E. Degree Examination, Jan./Feb. 2021

Basic Thermodynamics

Time: 3 hrs.

Max. Marks: 100

*Note: 1. Answer any FIVE full questions, choosing ONE full question from each module.**2. Use of Thermodynamic Data hand book and steam table is permitted.**3. Assume missing data suitably.*Module-1

- 1 a. Differentiate between :
- Microscopic and Macroscopic view point.
 - Open system and closed system.
 - Intensive and Extensive properties.
 - Mechanical and Thermal equilibrium. (10 Marks)
- b. An Engine cylinder has a piston of Area 0.12 m^2 and contains gas at a pressure of 1.5 MPa. The gas expands according to a process which is represented by a straight line on P-V diagram. The final pressure is 0.15 MPa. Calculate the work done by the gas on the piston if the piston stroke is 0.30 m. (10 Marks)

OR

- 2 a. State the definition of work in terms of thermodynamics and mechanics. (08 Marks)
- b. The temperature t on a certain Celsius thermometric scale is given by means of a property through a relation $t = a \ln p + b$ where a and b are constants and p is the property of the fluid. If at the ice point and steam points the values of p are found to be 4 and 20 respectively, what will be the temperature reading corresponding to a reading of $p = 16$? (12 Marks)

Module-2

- 3 a. Write an expression for displacement work for the following process with P-V diagram:
- Constant pressure
 - Constant volume
 - Constant temperature
 - Polytropic process (10 Marks)
- b. A fluid is confined in a cylinder by a spring loaded, frictionless piston so that the pressure in the fluid is a linear function of the volume ($P = a + bV$). The internal energy of the fluid is given by the following equation $U = 34 + 3.15 PV$, where U is in KJ, P in KPa and V in cubic metre. If the fluid changes from an initial state of 170 KPa, 0.03 m^3 to a final state of 400 KPa, 0.06 m^3 , with no work other than that done on the piston, find the direction and magnitude of the work and heat transfer. (10 Marks)

OR

- 4 a. With a neat sketch, explain the Joules experiment to investigate the equivalence of heat and work. State the first law of thermodynamics for a closed system undergoing a cycle and a change of state. (10 Marks)
- b. Briefly describe the different modes of energy and prove that energy is a property of the system. (10 Marks)

Module-3

- 5 a. What is a control volume and a control surface? With usual notation derive steady flow energy equation of 1st law of thermodynamics for an open system. (10 Marks)
- b. What do you understand by Qualitative difference between heat and work? List out any four similarities and differences between work and heat. (10 Marks)

OR

- 6 a. Briefly describe the following:
- (i) Kelvin Plank's II law of thermodynamics.
 - (ii) Clausius second law of thermodynamics.
 - (iii) Cyclic heat engine.
 - (iv) Thermal energy reservoirs. (10 Marks)
- b. Air flows steadily at the rate of 0.5 kg/s through an air compressor, entering at 7 m/s velocity, 100 KPa pressure and $0.95 \text{ m}^3/\text{kg}$ volume and leaving at 5 m/s, 700 KPa and $0.19 \text{ m}^3/\text{kg}$. The internal energy of the air leaving is 90 KJ/kg greater than that of the air entering. Cooling water in the compressor jacket absorbs heat from the air at the rate of 58 kW:
- (i) Compute the rate of shaft work input to the air in kW.
 - (ii) Find the ratio of the inlet pipe diameter to outlet pipe diameter. (10 Marks)

Module-4

- 7 a. With the help of a P-V and T-S diagrams derive an expression for the air standard efficiency of a diesel cycle. (10 Marks)
- b. An engine working on the otto cycle is supplied with air at 0.1 MPa, 35°C. The compression ratio is 8. Heat supplied is 2100 KJ/kg. Calculate
- (i) The maximum pressure and temperature of the cycle.
 - (ii) The cycle efficiency and
 - (iii) The mean effective pressure.
- (For air $C_p = 1.005$, $C_v = 0.718$ and $R = 0.287 \text{ KJ/kgK}$) (10 Marks)

OR

- 8 a. With P-V and T-S diagrams compare Otto diesel and dual cycles for,
- (i) The same compression ratio and heat rejection.
 - (ii) The same maximum pressure and temperature, the heat rejection being also the same. (10 Marks)
- b. With P-V and T-S diagrams derive an expression for thermal efficiency of a constant pressure closed cycle gas turbine in terms of Pressure Ratio (R_p). (10 Marks)

Module-5

- 9 a. A four cylinder petrol engine has a rated output of 52 kW at 2000 rpm. A Morse test is carried out and the brake torque readings are 177, 170, 168 and 174 N-m respectively. For normal running at this speed, the BSFC is 0.25 kg/kw-h and the calorific value of the fuel used is 42500 KJ/kg. Calculate the mechanical and brake thermal efficiency. (10 Marks)
- b. Express the Vander Waals constant in terms of critical properties. (10 Marks)

OR

- 10 a. Explain the Willian's line method and Morse test for determining the frictional power. (10 Marks)
- b. 1 kg of propane (C_3H_8) is at a pressure of 7 MPa and a temperature of 150°C. The critical properties of propane are $P_c = 4.26 \text{ MPa}$, $T_c = 370 \text{ K}$ and $V_c = 0.00454 \text{ m}^3/\text{kg}$ compressibility factor = 0.54. Calculate :
- (i) The reduced pressure, volume and temperature.
 - (ii) Specific volume of propane using ideal gas equation. (10 Marks)

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