

# CBCS SCHEME

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18SM33

## Third Semester B.Tech. Degree Examination, Feb./Mar. 2022 Engineering Thermodynamics

Time: 3 hrs.

Max. Marks: 100

**Note: 1. Answer any FIVE full questions, choosing ONE full question from each module.  
2. Use of thermodynamic data hand book is permitted.**

### Module-1

- 1 a. Explain with examples how thermodynamics play an important role in engineering and science. (08 Marks)
- b. Define the following with example:  
(i) Control volume  
(ii) Intensive and extensive property (06 Marks)
- c. The e.m.f. in a thermocouple with the test junction at  $t^{\circ}\text{C}$  on gas thermometer scale and reference junction at ice point is given by  $e = 0.20 t - 5 \times 10^{-4} t^2$  mV. The voltmeter is calibrated at ice and steam points. What will this thermometer read in a place where the gas thermometer reads  $50^{\circ}\text{C}$ ? (06 Marks)

OR

- 2 a. With P-T diagram, explain different regions of water that takes place. (06 Marks)
- b. Explain the following: (i) Compressibility factor (ii) Compressibility chart (06 Marks)
- c. A gas mixture consists 3 mole  $\text{O}_2$ , 5 mole  $\text{N}_2$  and 2 mole  $\text{CO}_2$ . Determine:  
(i) The mole fraction  
(ii) The mass fraction of each component  
(iii) Apparent molecular weight of mixture  
(iv) Apparent gas constant (08 Marks)

### Module-2

- 3 a. Differentiate between thermodynamic work and heat. (06 Marks)
- b. With a diagram, explain pdv-work. (06 Marks)
- c. If a gas of volume  $6000 \text{ cm}^3$  and at a pressure of 100 kPa is compressed quasistatically according to  $PV^2 = C$  until the volume becomes  $2000 \text{ cm}^3$ . Determine final pressure and work transfer. (08 Marks)

OR

- 4 a. State Joules experiment 1<sup>st</sup> law of thermodynamics. (06 Marks)
- b. Define specific heat at constant pressure and constant volume. (06 Marks)
- c. Steam is supplied to a 73.6 KW turbine with an enthalpy of 2800 kJ/kg velocity of 110 m/s. The steam leaves the turbine with a velocity of 300 m/s and an enthalpy of 2000 kJ/kg. Heat lost from the turbine to the surroundings is 20 kJ/kg. The change in potential energy is negligible. Determine: (i) The shaft work output (ii) Steam flow rate in kg/hr (08 Marks)

### Module-3

- 5 a. Define Kelvin Planck and Clausius statements. (06 Marks)
- b. Explain about the thermodynamic temperature scale. (06 Marks)
- c. A domestic refrigerator maintains a temperature of  $-12^{\circ}\text{C}$ . The ambient air temperature is  $35^{\circ}\text{C}$  if heat leaks to the freezer at the continuous rate of 2 kJ/sec, determine the maximum COP of the refrigerator and least power required to drive the refrigerant. (08 Marks)

OR

- 6 a. State and prove Clausius of inequality. (06 Marks)  
 b. Using Tds relation, find the entropy change for an ideal gas. (08 Marks)  
 c. Find the entropy change of 5 kg of a perfect gas whose temperature varies from 150°C to 200°C during constant volume process the specific heat varies linearly with absolute temperature and is represented by the relation  $C_V = 0.45 + 0.009 T$  kJ/kg°K. (06 Marks)

**Module-4**

- 7 a. With P-V and T-S diagram, derive efficiency of a diesel cycle in terms of compression ratio and cut-off ratio. (08 Marks)  
 b. Differentiate between open cycle and closed cycle gas turbine. (06 Marks)  
 c. A certain quantity of air at a pressure of 1 bar and temperature 70°C is compressed adiabatically to 7 bar in Otto cycle and 460 kJ of heat is added per kg of air, calculate:  
 (i) Compression ratio  
 (ii) Temperature at the end of compression  
 (iii) Temperature at the end of heat addition (06 Marks)

OR

- 8 a. With line and T-S diagram, explain Reheat Rankine Cycle and deduce the formula for efficiency. (06 Marks)  
 b. What are the effects of temperature and pressure on Rankine cycle? (06 Marks)  
 c. Steam at 20 bar, 360°C is expanded in a steam turbine to a pressure of 0.08 bar. It then enters a condenser. Where it is condensed to saturated liquid water. Assuming the turbine and feed pump efficiency as 60% and 90% respectively, determine per kg of steam, the net work, the heat transferred to working fluid and the Rankine cycle efficiency. (08 Marks)

**Module-5**

- 9 a. With a neat sketch, explain vapour compression refrigeration system. (10 Marks)  
 b. Define the following:  
 (i) COP (ii) Refrigeration effect (iii) Capacity of refrigerator  
 (iv) Mass flow rate of refrigerant (v) Power required (10 Marks)

OR

- 10 a. With a neat sketch, explain Summer Air Conditioning Systems. (08 Marks)  
 b. Define the following: (i) DBT (ii) WBT (iii) Specific humidity (06 Marks)  
 c. It is required to design an air conditioning plant for a office room with the following conditions:  
 Outdoor conditions: 14°C DBT and 10°C WBT  
 Required conditions : 20°C DBT and 60% RH  
 Air circulation : 0.3 m<sup>3</sup>/min/person  
 Seating capacity : 60  
 Condition is achieved first by heating and then by adiabatic humidifying. Determine:  
 (i) Heating capacity of coil in KW  
 (ii) Capacity of humidifier (06 Marks)

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